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22, Hydrocarbon Com
pounds(continued)

14. Circle the letter of
each condensed

structural formula for
pentane. a. C_5H_{12}

b. $CH_3CH_2CH_2$

CH_2CH_3 c. CH_3

$(CH_2)_3CH_3$ d. $C—$

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C — C — C — C 15. The IUPAC system uses _____ to show the number of carbon atoms

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744 Chapter 21 •
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221.11.1 Introduction
to Hydrocarbons

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MAIN Idea

Hydrocarbons are carbon-containing organic compounds that provide a source of energy and raw materials. Real-World Reading Link If you have ridden in a car or a bus, you have used hydrocarbons. The gasoline and diesel fuel that are used in cars, trucks,

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Key 1. Answer the following. a. What is the general formula for a noncyclic alkane? C_nH_{2n+2} b. If a noncyclic contains 15 carbon atoms, how many hydrogen atoms does it contain? 32 2. Draw the line bond structures of following types of hydrocarbons using four carbon atoms: a. two

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Exercise 1c 1 Sean –
He was named after
the actor Sean
Connery, who played
James Bond in the
60s.

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What are
Hydrocarbons?

Hydrocarbons are organic compounds containing only hydrogen and carbon atoms. For example, the gas used in gas stoves and bunsen burners can be either propane (C_3H_8) or butane (C_4H_{10}).

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hydrocarbons burn

'cleanly' to form

water and carbon

dioxide in the

presence of sufficient

oxygen.

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Key Concepts 22.1

Hydrocarbons •

Because carbon has four valence

electrons, carbon

atoms always form

four covalent bonds. •

The carbon atoms in

an alkane can be

arranged in a straight

chain or in a chain

that has branches. •

Molecules of

hydrocarbons, such

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as alkanes, are
nonpolar molecules.
22.2 Unsaturated
Hydrocarbons

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one formula with all five carbon atoms in a horizontal line and the other with four carbon atoms in a line, with a CH₃ group extending down from the first attached to the third carbon atom.

12.E: Organic
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Day 1: Alkanes - You must know how to name and draw straight chain and branched alkanes.

Homework: Finish the Straight Chain Alkane practice at the end of the notes. GOALS:

Naming, molecular formula, structural formula, condensed formula, and carbon skeletal formula. I

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