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Date Problem Solving continued Sample Problem 4 How many boron atoms

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are there in 2.00 g of boron? Solution ANALYZE What is given in the problem? mass of boron in grams What are you asked to find? number of boron atoms PLAN What steps are needed to convert from grams of B to ...

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$g(x) = 6(0.2x) + 5$. Select the correct choice below, and fill in the answer box to complete your choice. A. The graph of $g(x)$ is translated unit (s) to the left compared to the graph of $f(x)$. B. The graph of $g(x)$ is translated unit (s) down compared to graph of $f(x)$.

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Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about $\frac{1}{5}$ of the molar mass of boron. Therefore, 2.00 g boron will contain about $\frac{1}{5}$ of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

Skills Worksheet Problem Solving

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